

Note Taking Guide Episode 303 Answers

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3-16 - Note Taking Guide Ep 303-Pt 1

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Note Taking Guide: Episode 303 Name Aditi Acharya Bohr's Energy Levels • Electrons in certain energy levels • Lower energy levels: closer to nucleus • Higher energy levels: further from nucleus • Ground State: electron in lowest energy level possible Excited Atom • Atom has absorbed energy • Excited state is unstable • Atom soon emits same amount of energy absorbed • energy seen as visible light Wave Description of Light Wavelength (λ): distance between corresponding points ...

note_ep303.doc - $F = 5 \cdot 10^{14} \text{ 1/s} = c = \cdot f \cdot 3 \cdot 10^8 / 5 \cdot 10^{14} \dots$

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