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Atom. Smallest particle of an element that retains the properties of that element during chemical reaction.

Democritus.

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Atomic Structure | Class 9 Chapter 4 Structure of Atom ...

CHAPTER 4 TEST: Atoms, Atomic Theory and Atomic Structure Matching. A. Bohr B. Democritus C. Rutherford D. Dalton E. Thomson F. Schrodinger _____ 1. Greek thinker; called nature's basic particle an atom, based on the Greek word "atomos" which means "indivisible". Did

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not have evidence that atoms existed.

____2. <http://www.brooklyn.k12.oh.us/userfiles/75/Classes/4711/NON-HONORS%20Chapter%204%20Test%202010.pdf>.

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that element. Democritus lived from 460 B.C. - 370 B.C. and was among the first to suggest the idea of atoms. Chapter 4 (Atomic Structure) Test - Chapter 4(Atomic ...

Chapter 4 Atomic Structure Test A Answers

Chapter 4: Atom (Test Answers) STUDY.

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Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. chxistina. Key Concepts: Terms in this set (33) Dalton's atomic theory was accepted because a. Dalton showed how new molecules are formed. b. Dalton invented the electron microscope. c. Democritus said that it was correct.

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Studying Atoms 4.2: The Structure of an
Atom 4.3: Modern Atomic Theory

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Section 4.2 Structure of the Nuclear

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Atom One change to Dalton's atomic theory is that atoms are divisible into subatomic particles: Electrons, protons, and neutrons are examples of these fundamental particles There are many other types of particles, but we will study these three

Chapter 4 Atomic Structure - Henry

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identify the experiments that led to the development of the nuclear model of atomic structure. You will describe the structure of the atom and differentiate among the subatomic particles that comprise it. You will explain the relationship between nuclear stability and radioactivity. Why It's Important

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Atomic Structure (Chapter Test A)

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emily_reidt. Terms in this set (21) the

total number of protons and neutrons in

the nucleus of an atom. mass number.

the weighted average mass of the atoms

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in a naturally occurring sample of an element.

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two species which can be termed isobars are: (i) $^{201}_{60}\text{X}$ (ii) $^{200}_{61}\text{X}$ (iii) $^{200}_{58}\text{X}$ (iv) $^{203}_{60}\text{X}$ (a) (i) and (ii) (b) (ii) and (iii) (c) (i) and (iii)...

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Test: Wednesday, Oct. 21 1 Review

Chapter 4 Atomic Structure Test

Wednesday, Oct. 21 2. Daltons Atomic

Theory and other important scientists ;

Rutherford Goldstein ; Thomson Gell-

Mann ; Bohr Chadwick ; Schrodinger

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Democritus ; Millikan ; 3. An element has an atomic number of 76. What is

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from this chapter. A B; atomic mass unit: defined as the mass of a carbon -12 atom: nucleus: central core of an atom, which contains most of the atom's mass: group: a vertical column of elements ...

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