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Chapter 15 Water And Aqueous

Chapter 15 Water and Aqueous Systems 159 SECTION 15.1 WATER AND ITS PROPERTIES (pages 445–449) This section describes the properties of water in the liquid and solid states and explains how hydrogen bonding affects the surface tension and vapor pressure of water. Water in the Liquid State (pages 445–447) 1.

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SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449)

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Chapter 15 Water and Aqueous Systems.
Chapter 15 "Water and Aqueous
Systems". The Water Molecule: a
Review. • Water is a simple tri-atomic
molecule, H₂O. • Each O-H bond is
highly polar, because of the high
electronegativity of the oxygen (N, O, F,
and Cl have high values) • bond angle of
water = 105°.

Chapter 15 Water and Aqueous Systems

Chemistry, Chapter 15, Water and
Aqueous Systems. surface tension.
surfactant. aqueous solution. solvent.

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the inward force or pull that tends to minimize the surface ar.... any substance that interferes with hydrogen bonding between wa.... is water that contains dissolved substances.

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Chapter 15 "Water and Aqueous Systems". use these activities to yourself study the vocabulary and major concepts presented in this chapter. heterogeneous mixture containing particles that are small enough to remain dispersed in the solvent and do not separate on standing.

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Section 15.1 Water and its

Properties OBJECTIVES: • Explain the high

surface tension and low vapor pressure

of water in terms of the structure of the

water molecule and hydrogen bonding. •

Chemistry - Chp 15 - Water and Aqueous Systems - Notes

Chapter 15 - Water and Aqueous

Systems - 15.3 Heterogeneous Aqueous

Systems - 15.3 Lesson Check - Page 507:

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21 Answer Colloids cannot be separated through filtering because their particles are smaller than those of suspensions and they do not settle out over time.

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Title: Chapter 15 Review Water and Aqueous Systems 1 Chapter 15 Review Water and Aqueous Systems. Pre-AP Chemistry ; Charles Page High School ; Stephen L. Cotton; 2 Chapter 15 Review. Surface tension is the _____. How does the surface tension of water compare with the surface tensions of most other liquids? Which type of mixture(s) exhibit

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OBJECTIVES: -Explain the high surface tension and low vapor pressure of water in terms of the structure of the water molecule and hydrogen bonding. 3.

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activities ... a relatively strong intermolecular force responsible for water's high surface tension:
suspension: a mixture from which particles settle out upon standing:
aqueous solution: a solution in which the solvent is water: solvent: the ...

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The structure of ice is a regular open framework of water molecules arranged like a honeycomb.

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SECTION 15.1 WATER AND ITS
PROPERTIES (pages 445-449) CHAPTER
15, Water and Aqueous
Systems(continued) 6 Circle the letter
next to each sentence that describes a
result of the surface tension of water a
In a full glass of water, the water surface

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seems to bulge over the rim of the glass
b Water beads up into

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