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Section 10.3 - Percent Composition and Chemical Formulas.

The percent by mass

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(percent composition) of an element in a compound is the number of grams of the element divided by the mass in grams of the compound multiplied by 100%. % mass of element = $\frac{\text{mass of element}}{\text{mass of compound}} \times 100$.

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**CHAPTER 2/CHAPTER
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SI unit representing
 6.02×10^{23}
representative
particles of a
substance. the
temperature and
pressure at which one
mole of gas occupies a
volume of 22.4 L. equal
volumes of gases at
the same temperature
and pressure contain

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equal numbers of
particles.

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In this lesson, we'll explore what an enzyme is, what an enzymatic reaction is and what happens to the rate of a reaction if you add more enzymes
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10 1 kg 103 g 368

Chapter 10 Chemical
Calculations and
Chemical Equations.

10.1 Equation

Stoichiometry 369 The
ratio of moles of P 4O
10 to moles of P (which
came from the

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subscripts in the
chemical formula, P 40
10) provided the key
conversion factor that
allowed us to convert

Chapter 10 Chemical Calculations and equations

Chemical Quantities

THE MOLE AND

QUANTIFYING MATTER

10.1 The Mole: A

Measurement of Matter

Essential

Understanding The

mole represents a

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large number of very small particles.

Reading Strategy

Frayer Model The

Frayer Model is a vocabulary

development tool. The center of the

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Chapter 10 Chemical
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10.1 THE MOLE: A
MEASUREMENT OF
MATTER (pages
287-296) This section

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defines the mole and explains how the mole is used to measure matter. It also teaches you how to calculate the mass of a mole of any substance.

Measuring Matter
(pages 287-289)

SECTION 10.1 THE MOLE: A MEASUREMENT OF MATTER (pages 287-296)

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equations worksheet is suitable for 10th 12th grade in this chemical equations worksheet students write and balance chemical reactions they complete 98 short answer and problem solving questions on

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- However, they can all be written by chemical equations that

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Chemists use to describe chemical reactions. • In every chemical reaction, atoms are rearranged to give new substances. -Just like following a recipe, certain ingredients are combined (and often heated) to form something new.

Chapter Seven 7.1

-Equations for
Chemical Reactions

Chemical Reactions

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percent

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